While these results demonstrate the potential of allyl sulfones as substrates in molybdenum-catalyzed allylic alkylations, it should be noted that they do have diminished reactivity compared to the corresponding allyl acetates. Agents which coordinate to molybdenum, such as acetonitrile or dioxane solvent and BSA¹⁴ base, inhibit reaction. There is also a significant $S_N 1$ component to the reaction. Tertiary sulfones react faster than the corresponding primary sulfones. Thus, the sulfone isomeric to 3, prenyl phenyl sulfone, required 4 times as long to produce the same products. The regiochemistry exhibited here is best analyzed as a delicate balance between steric and electronic factors. Sterically demanding nucleophiles attack the intermediate π -allylmolybdenum complex at the least substituted position. Small nucleophiles such as dimethyl malonate, however, react under electronic control and attack at the site of greatest electron deficiency, which is the more substituted allylic terminus. Thus molybdenum catalysis can provide a unique entry into systems containing a quaternary carbon center.

(14) BSA = O,N-bis(trimethylsilyl)acetamide.

Acknowledgment. We wish to thank the National Science Foundation and the National Institutes of Health, General Medical Sciences Institute, for their generous support of our programs and the Fannie and John Hertz Foundation for a fellowship for C.A.M. We thank Pressure Chemical Co. for a gift of molybdenum hexacarbonyl.

Registry No. 3, 72863-20-8; 4, 74866-35-6; 5, 43219-18-7; 6, 97072-44-1; 7, 124535-79-1; 8, 124535-80-4; 9, 124535-81-5; 10, 124535-82-6; 11, 124535-83-7; 12, 124535-80-4; 9, 124535-81-5; 10, 124535-82-6; 15 (n = 2), 124535-85-9; 16, 124535-86-0; 17, 124535-87-1; 18, 124535-88-2; 19, 124535-89-3; 20, 124535-90-6; 21, 124535-91-7; 22 (isomer 1), 124561-56-4; 22, 97072-38-3; 23, 124535-92-8; 24, 124535-93-9; 25, 124535-94-0; NaCH(CO₂Me)₂, 18424-76-5; Mo(CO)₆, 13939-06-5; H₂C=CHCH₂SO₂Ph, 16212-05-8; Br(CH₂)₄Br, 110-52-1; Pd(PPh₃)₄, 14221-01-3; I(CH₂)₂O(C-H₂)₂I, 34270-90-1; Cl(CH₂)₂Cl, 107-06-2; Br(CH₂)₃Br, 109-64-8; H₂C=CHCH(CO₂Me)₂, 40637-56-7; (E)-CH₃CH=CHCH₂SO₂Ph, 72863-24-2; (Z)-MeOCH=CHCH₂SO₂Ph, 124535-95-1; I(CH₂)₃Cl, 6940-76-7; 2-(carbomethoxy)cyclopentanone, 10472-24-9.

Supplementary Material Available: Characterization data for 6, 7, 9–14, 16, 17, 20, 23–25 and sample procedure for alkylation (3 pages). Ordering information is given on any current masthead page.

Synthesis of 16-Membered Macrolide Aglycons, Carbonolide A, Leuconolides, and Maridonolides, via Carbonolide B Type Compounds by Virtue of Completely Stereoselective Epoxidation and Reduction Based on the Conformational Control of Macrolide Rings with Protecting Groups¹

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Summary: Carbonolide B type compounds were converted to seven typical 16-membered macrolide aglycons, carbonolide A, EOP aglycon, leuconolide A_1 and A_3 , midecanolide A_1 , and maridonolide II and I, by virtue of completely stereoselective epoxidation and/or reduction based on the conformational control of macrolide rings with protecting groups. NOE and NOESY measurements and MMP2-CONFLEX2 calculations were employed to predict the conformation of the carbonolide B type compounds.

The 16-membered aglycons of the largest group of macrolide antibiotics represented by carbomycins, leucomycins, and maridomycins are classified into four types, 1, 2, 3, and 4, according to their oxidation levels² and, except for 1, most of these remain unsynthesized.³ We recently reported the stereoselective total synthesis of typical macrolide aglycons⁴ by virtue of the MPM (methoxyphenylmethyl) protection of hydroxy functions⁵ and some stereocontrolled reactions in acyclic systems. This methodology, together with stereoselective epoxidation and reduction on 16-membered lactone rings,⁶ is now extended to the first completely stereoselective synthesis of 16-

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membered macrolide aglycons: carbonolide A (2b),⁸ leuconolide A₁ (3a)⁹ and A₃ (josanolide) (3b),¹⁰ midecanolide

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9: R¹=H, R²=OH 10: R1=OH, R2=H

^a (a) MCPBA (4 equiv), NaHCO₃, CH₂Cl₂, 20 °C; (b) DDQ, CH₂Cl₂-H₂O (20:1); (c) Bu₄NBH₄ (2 mol), MeOH, room temperature, 1.5 h.

 A_1 (3c),¹¹ and maridonolide II (4b) and I (4c),¹² etc.

When the dienone (6), synthesized as an intermediate to carbonolide B (1b) from D-glucose and L-malic acid via 5^{13} was oxidized with MCPBA in the presence of NaHCO₃, completely regio- and stereoselective epoxidation occurred to give only the expected β -epoxide (7), 6,7a,b which was readily converted to carbonolide A (2b) by removal of the MPM group.^{5,14}

In order to synthesize the leuconolides and maridonolides, the reduction of 6 and 7 was next examined. When 6 was treated with n-Bu₄NBH₄ in MeOH at room temperature, a 1:1.8 mixture of the desired 9R alcohol (9) and its 9S isomer (10) was obtained.^{7c} Reduction of 7 under

(13) Nakajima, N.; Uoto, K.; Yonemitsu, O. Heterocycles, in press. (14) EOP aglycon $(2c)^3$ was also synthesized from 8 in the same way.

the same conditions gave mainly an undesired 9S alcohol with 22:1 selectivity. This disappointing selectivity is clearly explained in terms of an unfavorable type A conformation of the dienone portion (Figure 1) on the basis of NOE and NOESY data, which is confirmed by the MMP2 calculation combined with a systematic structure generation algorithm (CONFLEX2).^{15,16} As can be seen from the computer drawing (though for 12, Figure 1), upon reduction of 6, the reducing agent can attack only from the left-side (arrow) of the A conformer to give 10, whereas the B conformer can give the expected 9 by the right-side attack of the reductant.¹⁷ If the population of the A conformer is lowered by variation of the protection pattern of C3, C5, and C6", hydroxy groups, a higher yield of the desired 9R (α) alcohol can be expected. According to NOE and NOESY measurements, the 3,5-acetonide (5) consists of interconvertible A, B, C, and A' conformers whose calculated populations are 67.9, 10.5, 10.5, and 8.8%, re-

⁽⁶⁾ There are a few precedents of the epoxidation and reduction of similar 16-membered dienones in the tylonolide series. 3-Deoxy-rosaranolide⁷ and rosaranolide⁷ were stereoselectively synthesized by the epoxidation of the corresponding dienones with MCPBA, but the reduction of a tylonolide derivative mainly gave an undesired 9S-hydroxy compound. $^{7\mathrm{c}}$

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⁽¹⁶⁾ Compound 11 in the same system was calculated to exist mainly in the A conformer (9,10-s-cis, 11,12-s-trans) (97.9%) together with the B conformer (9,10-s-cis, 11,12-s-cis) (2.0%).

⁽¹⁷⁾ Since in the A conformer the dienone of 6 is almost at right angles to the 16-membered ring plane, the inside of the ring (the right-side of the computer drawing) is completely blocked. The 1:1.8 ratio of 9 and 10 on the reduction of 6 in spite of the 1:49 population ratio of the B and A conformers (interconvertible at room temperature according to the NMR spectrum of 6) clearly shows that the B conformer was reduced much faster than the A conformer, whose reduction was subject to steric hinderance by the C8 methyl group [the left-side of the B conformer is sterically hindered by the C7 (methylene) group]. Reliable calculation for the epoxide (7) are still impossible, but the population of the B conformer must be nearly nil.



Δ





в

(omission of hydrogens)

Figure 1.

Scheme II^a



^a (d) NaBH₄, MeOH, 0 °C, 5 min; (e) (ClCH₂CO)₂O (3 equiv), DMAP (1 equiv), py; (f) (COCl)₂ (2 equiv), DMSO (4 equiv), CH₂Cl₂, Et₃N (6 equiv), $-78 \rightarrow 0$ °C; then 1 N HCl-THF (1:5); (g) K₂CO₃ (3 equiv), MeOH, 0 °C, 5 min; (h) Ac₂O, Et₃N, DMAP, CH₂Cl₂; (i) CSA (0.1 equiv), MeOH, 20 °C, 10 min; (j) Me(MeO)C=CH₂ (5 equiv), PPTS (0.1 equiv), CH₂Cl₂, 20 °C.

spectively.¹⁸ Among them the B, C, and A' conformers (total population: 29.8%) should be reduced to afford the desired 9R alcohol.

When 5 was treated with $NaBH_4$ in MeOH at 0 °C, a rapid stereoselective reduction occurred to give the expected alcohol (13) in almost quantitative yield. Protection with a chloroacetyl group followed by DDQ oxidation gave

the primary alcohol (14), which was oxidized to the aldehyde. The acetonide group was then hydrolyzed to give the hemiacetal (15), and finally an alkaline hydrolysis gave leuconolide A_1 (3a). Acetylation of 15 gave the diacetate (16), which was easily converted to leuconolide A_3 (3b).¹⁹ Similarly, midecanolide A_1 (3c) was synthesized from 15.

⁽¹⁸⁾ Data calculated for 12. C: 9,10-s-trans, 11,12-s-trans. A': 9,10s-cis, 11,12-s-trans; the dienone portion inside out.

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Oxidation of 5 with MCPBA gave the epoxide (17) in poor yield, but a better result was obtained via three conventional reactions. When 17 was reduced with NaB- H_4 , the expected 9R alcohol (18) was obtained in quantitative yield. This (18) was converted to maridonolide II (4b) via 19-21 as described for 3b. Compound 4b was also obtained from 16 via only two reactions: MCPBA oxidation and hydrolysis with K_2CO_3 . Similarly, maridonolide I (4c) was synthesized from 5.

The aglycons 2b,c, 3a-c, and 4b,c thus synthesized were identical in all respects (NMR, MS, IR, $[\alpha]_D$) with those derived from natural antibiotics.²⁰

Acknowledgment. We thank Yamanouchi Pharmaceutical Co., Dr. K. Hosoi, Meiji Seika, Ltd., and Dr. T. Kishi, Takeda Chemical Industries, Ltd., for kind supplies of josamycin, midecamycin A₁, and maridomycin III, respectively.

Supplementary Material Available: Physical data for compounds 2b,c, 3a-c, 4b,c, 13, 15, 17, and 18 and calculation data (MMP2-CONFLEX2) for 11 and 12 (26 pages). Ordering information is given on any current masthead page.

(20) These aglycons (isomeric mixtures with respect to the hemiacetal positions) except for $3b^{10,19}$ have never been reported.

Biosynthesis of Furaquinocins A and B

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Summary: The biosynthetic pathway to furaquinocins was investigated by means of incorporation experiments with $[1-^{13}C]$ acetate, $[1,2-^{13}C_2]$ acetate, and $[methyl-^{13}C]$ -Lmethionine. Furaquinocins are derived from a pentaketide, two mevalonates, and two C_1 units from Lmethionine.

The furaquinocins A(1) and B(2) comprise a novel polyketide antibiotic complex produced by Streptomyces sp. KO-3988, which exhibit cytotoxic activity against HeLa S₃ cells but no antimicrobial activity.¹ Extensive spectroscopic analyses have demonstrated that 1 and 2 possess a unique structure, which consists of a naphtho [1,2-b]furan-6,9-dione chromophore with an isoprenoid-like side chain.²



1
$$R_1 = H$$
, $R_2 = OH$
2 $R_1 = OH$, $R_2 = H$

In the present report, we describe studies on the biosynthetic pathway to furaquinocins A and B by means of feeding experiments using ¹³C-labeled precursors.³

Assignment of the ¹³C NMR signals of 1 and 2 was fully established on the basis of ¹H-¹³C COSY along with LSPD

Table I. ¹³C NMR Chemical Shifts, Enrichment Ratio of [1-¹³C]Acetate-Labeled Furaquinocins, and J_{CC} (in Hz) of [1,2-13C2]Acetate-Labeled Furaquinocins^a

		1			2	
		enrichmt			enrichmt	
carbon	δ	ratio ^b	J	δ	ratio ^b	J
2	88.9	1.0	38	88.9	1.0	38
Me-2	16.1	5.1	38	16.1	5.2	38
3	52.8	3.3	35	52.4	3.2	34
Me-3	18.9	1.0	35	18.9	0.8	34
3a	124.6	0.4	61, 69	124.5	0.2	61, 70
4	158.9	6.9	69, 67	158.4	5.8	70, 67
5	111.0	0.8	67,66	110.7	1.0	67, 66
5a	134.0	4.2	66, 54	134.1	4.3	66, 54
6	180.8	0.4	54, 57	180.7	0.3	54, 57
7	156.9	5.3	57, 74	156.9	3.8	57, 74
MeO-7	60.6	0.8	с	60.7	0.4	с
8	133.6	0.8	74, 52	133.7	0.6	74, 52
Me-8	9.3	1.0	с	9.3	0.8	с
9	183.8	4.5	52, 60	183.7	4.4	52, 60
9a	108.8	0.3	60, 73	109.2	0.2	60, 73
9b	160.6	4.0	73, 61	160.4	4.8	73, 61
10	71.4	0.8	sd	73.0	1.2	sd
11	32.4	6.0	43	31.9	7.7	44
12	124.9	0.7	43	120.1	0.9	44
13	138.3	4.0	46	140.0	5.0	43
14	23.2	1.0	s^d	68.0	0.7	s ^d
15	61.4	0.8	46	14.3	0.8	43

^a Each sample was dissolved in CDCl₃ and chemical shifts were shown with reference to $CDCl_3$ as 77.0 ppm. ^bEnrichment ratios were relative to the C-2 signal as 1.0. ^cSignal for the carbon not incorporated. ^dSignal was singlet, so the carbon had no coupling with others.

experiments² and are presented in Table I. The ${}^{13}C$ NMR spectra of furaquinocins A (1) and B (2) labeled with [1-¹³Clacetate revealed the enrichment of nine carbon signals (Me-2, C-3, C-4, C-5a, C-7, C-9, C-9b, C-11, and C-13). The intensity ratios of nonlabeled ones are also shown in Table I. In a feeding experiment with [1,2-¹³C₂]acetate, 20 carbons were found to be derived from acetate, of which 11 carbons (C-2, Me-3, C-3a, C-5, C-6, C-8, C-9a, C-10, C-12, C-14, and C-15) arose from C-2 of acetate. The remaining two carbons (MeO-7 and Me-8) were not derived from acetate units. The ${}^{13}C{}^{-13}C$ coupling constants (J_{CC}) of 1 and 2 labeled with $[1,2^{-13}C_2]$ acetate are given in Table I. The carbons in the naphthoquinone ring show two kinds of coupling with equal signal intensities. These coupling constants are coincident with those of the adjacent carbons.

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 (3) A seed culture of Streptomyces sp. KO-3988 was made by the method previously described.¹ The seed culture (1-2 mL) was inoculated into each 500-mL Sakaguchi flask containing 100 mL of medium combined 500 mL of the formation of the flask containing 100 mL of the flash containin posed of 2% starch, 1% soy bean meal, and 0.3% NaCl (pH 7.0 before sterilization). ¹⁸C-labeled precursors in 2 mL of aqueous solution ([1-¹³C]and $[1,2^{-13}C_2]$ acetate, 100 mg; [methyl-13C]-L-methionine, 50 mg) was added to each flask after 6 h of cultivation and the cultures were then incubated for 66 h at 27 °C. ¹³C-enriched furaquinocins (2-4 mg) were isolated for the cultured broth (0.5-1 L) by solvent extraction, silica gel chromatography, and Sephadex LH-20 gel filtration as previously reported.¹ ¹⁸C NMR spectra were recorded on a Varian XL-400 spectrometer in deuteriochloroform.